Tandem Reactions

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Synthesis of Heterocycles through a Ruthenium-Catalyzed Tandem Ring-Closing Metathesis/Isomerization/N-Acyliminium Cyclization Sequence**

Erhad Ascic, Jakob F. Jensen, and Thomas E. Nielsen*

Olefin metathesis is an extremely powerful and general method for carbon-carbon bond formation in organic synthesis.^[1] For example, ruthenium alkylidene catalyzed metathesis has been widely used to construct a variety of alkenes for applications in chemistry, materials science, and chemical biology. A key asset of the metathesis process is the unique olefin functional group selectivity mediated by robust and well-defined catalytic systems.

Over the years, however, unexpected nonmetathetic reactions have been observed under metathesis conditions.^[2] Although these reactions typically are highly substrate dependent, associated with specific reaction conditions, and possibly caused by ill-defined metal-catalytic species, they represent a unique opportunity for the development of tandem processes.[3] It is well recognized that tandem reactions offer major advantages in the synthesis of valuable target compounds. In this context, metathesis mediated by ruthenium alkyli-

dene catalysts 1e and 1k (Grubbs first- and second-generation catalysts; Figure 1) has successfully been coupled to nonmetathetic transformations, such as double-bond isomerization, [4-6] hydrogenation, [7-9] cyclopropanation, [10] dihydroxylation, [11,12] keto-hydroxylation, [12] and Kharasch addition reactions.[13]

Only a few reports have dealt with the tandem ringclosing metathesis (RCM)/ double-bond isomerization. Notable works by the groups of Snapper^[4] and Schmidt^[5] have independently shown how cyclic allyl ethers can isomerize into 2,3-dihydropyrans. Schmidt and co-workers have also shown the beneficial effect of added hydride to favor the isomerization step. Inspired by the work of Fustero et al. (RCM/isomerization), [14] and Pérez-Castells and co-workers

[*] E. Ascic, Dr. J. F. Jensen, Dr. T. E. Nielsen Department of Chemistry, Technical University of Denmark 2800 Kgs. Lyngby (Denmark) Fax: (+45) 4593-3968 E-mail: ten@kemi.dtu.dk Homepage: http://www.thomasenielsen.org

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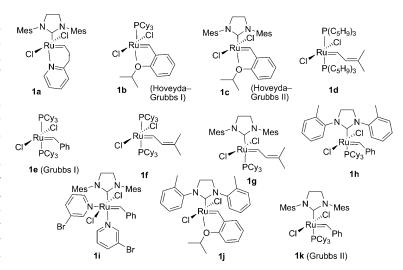


Figure 1. Ruthenium alkylidene catalysts commonly used in metathesis. Cy = cyclohexyl, Mes = 2,4,6-trimethylphenyl.

(RCM/isomerization/cyclopropanation), [15] we speculated that enamides generated in the event of a RCM/isomerization sequence could be further isomerized into reactive Nacyliminium intermediates (Scheme 1).[16] The presence of a suitably tethered nucleophile could then bring about a second cyclization step.

Scheme 1. Tandem RCM/isomerization/N-acyliminium cyclization.

Initial investigations were focused on substrate 2 (see, Table 1), which contains an indole moiety as a potentially reactive π nucleophile. The resulting product has a tetracyclic indolizinoindole core that is present in a range of pharmacologically interesting compounds, such as GPCR antagonists, [17] antibacterial, [18] and antiparasitic agents. [19] Access to enantiopure indolizinoindole derivatives has also been widely pursued in recent efforts in asymmetric catalysis. [20] We started out by screening ruthenium catalysts 1a-k (Figure 1), in toluene at reflux (Table 1). The reactions were generally very clean, as indicated by UPLC-MS and ¹H NMR spec-

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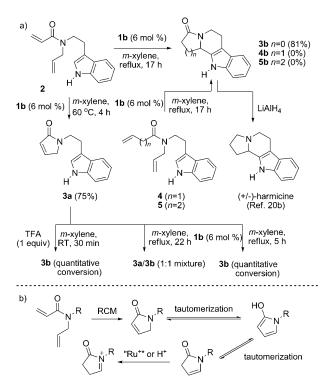


Table 1: Screening of catalysts for RCM/isomerization/N-acyliminium cyclization.

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	Entry	Catalyst	Solvent	2/3 a/3 b ^[a]
	1	1a	toluene	100:0:0
	2	1 b	toluene	0:2:98
	3	1 b	m-xylene	0:0:100
	4 ^[b]	1 b	m-xylene	0:5:95
	5	1 c	toluene	0:26:74
	6	1 c	m-xylene	0:0:100
	7 ^[b]	1 c	m-xylene	0:10:90
	8	1 d	toluene	83:3:14
	9	1 e	toluene	54:6:40
	10	1 f	toluene	36:0:64
	11	1 f	m-xylene	0:0:100
	12 ^[b]	1 f	m-xylene	38:0:62
	13	1 g	toluene	0:12:88
	14	1 g	m-xylene	0:0:100
	15 ^[b]	1 g	m-xylene	0:5:95
	16	1 h	toluene	61:4:35
	17	1i	toluene	25:38:37
	18	1 j	toluene	3:14:83
	19	1 j	m-xylene	0:0:100
	20 ^[b]	1 j	m-xylene	7:13:80
	21	1 k	toluene	0:70:30

[a] With the exception of entry 17, product mixtures were generally clean (>85% of 2/3a/3b in the reaction mixture as indicated by RP-HPLC analysis). [b] The reaction was carried out with 5 mol % of catalyst.

troscopy, and the proposed enamide intermediate was never detected.^[21] Notably, with 15 mol % of catalyst in m-xylene at reflux, several catalysts brought about clean conversion into the desired product **3b** (entries 3, 6, 11, 14, and 19). When the amount of catalyst was lowered to 5 mol %, the Hoveyda-Grubbs catalyst 1b gave the cleanest and highest conversion (95%) of 2 into 3b (entry 4). When the catalyst loading was raised to 6 mol %, the reaction was complete in 17 hours (Scheme 2a), and gave 3b in 81% yield. When 2 was subjected to the same reaction conditions at lower temperature (60°C), the metathesis product 3a could be isolated in 75% yield. After careful removal of trace amounts of ruthenium, $^{[22]}$ the conversion of 3a into 3b was investigated under various reaction conditions. When 3a was treated with catalyst 1b, the desired product formed quantitatively in less than 5 hours. The same experiment without catalyst led to a 1:1 mixture of 3a and 3b, thus indicating a thermal background reaction, but also unambiguously demonstrating the beneficial effect of the catalyst in the nonmetathetic part of the tandem sequence. The synthesis of 3b represents a formal total synthesis of the antiparasitic natural product harmicine. [19,20b] The homologous substrates 4 and 5 were not converted into the tetracycles 4b and 5b under the optimized conditions (although they still underwent RCM reactions) and TFA rapidly converted 3a into 3b (Scheme 2a); together these findings suggest to us that the nonmetathetic role of the ruthenium does not necessarily involve a ruthenium hydride intermediate, but somehow promotes favorable tautomeriza-



Scheme 2. a) Optimized reaction conditions, and b) proposed mechanism for formation of N-acyliminium intermediate. TFA = trifluoroacetic acid.

tion events during the isomerization (Scheme 2b). In additional experiments, we also noted that the conversion of 2 into **3b** can be catalyzed cleanly with **1b** in the presence of 1–2% TFA or BF₃·Et₂O (added at the beginning of the reaction) in m-xylene heated at reflux, thus shortening the reaction times to less than 1 hour but giving slightly lower yields (70-75%; Scheme 2a).

The reaction using the N-methylated indole 6 gave an increased yield of 86% (Scheme 3). The introduction of a substituent, as present in the tryptophan (8) and benzothienylalanine (10) derivatives, effectively directed the formation of the new stereocenter with excellent trans diastereoselectivity at the ring junction. The trimethoxybenzene derivative 12 also underwent the tandem reaction sequence to give the tetrahydroisoquinoline derivative **13** in good yield (64%). The methodology was also extended to an intermolecular variant, wherein indole acted as the nucleophile in the reaction with the N-acyliminium intermediate that was derived from 14, in good yield (57%). Under these reaction conditions, however, the reaction required the addition of 4 equivalents of indole. If 14 was instead treated with 1b at reflux for 1 hour, followed by the addition of indole (1 equiv) and TFA (1 equiv) and additionally reacted for 1 hour, then 15 could be isolated in 84% yield. The nucleophilicity of the aromatic ring is highly important, as evident for substrates 16, 18, and 20 (Scheme 4), which were not converted into the corresponding tricycles under reaction conditions similar to those used in Scheme 3. Ring-closing metathesis occurred smoothly for these substrates but further conversion was better mediated by the subsequent addition of 1-4 equiva-

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Scheme 3. Tandem RCM/isomerization/N-acyliminium reactions.

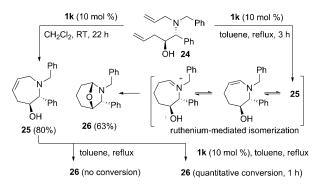
Scheme 4. RCM and acid-mediated isomerization/N-acyliminium cyclization.

lents of TFA to the reaction mixture. In this way, tricyclic compounds 17, 19 and 21 were obtained in good to excellent yields.

Furthermore, the extension of the methodology to heteroatom nucleophiles was briefly examined by using substrates **22a**–**c**, and, rewardingly, hemiaminals **23a**–**c** were formed in good yield and with excellent diastereoselectivity (Scheme 5).

 $\begin{tabular}{ll} {\it Scheme 5.} & Tandem RCM/some rization/N-acyliminium cyclization reactions of alcohols. \end{tabular}$

The tandem methodology presented herein, presumably involves the generation of an N-acyliminium species. Therefore, it was natural to also investigate whether N-alkyliminium ions could be formed during a similar tandem process. In a preliminary study on amino alcohol **24**, treatment with 10 mol% of the Grubbs second-generation catalyst **1k** resulted in a tandem metathesis/isomerization/cyclization sequence to give the bicyclic product **26** in good yield (Scheme 6). Notably, carefully purified cyclic alkene **25** does not convert into **26** under thermal conditions whereas the addition of **1k** rapidly effects the isomerization steps.



Scheme 6. Ruthenium-catalyzed tandem RCM/isomerization/N-acyliminium cyclization.

In summary, an efficient ruthenium-catalyzed tandem ring-closing metathesis/isomerization/N-acyliminium cyclization sequence has been developed. In this tandem process, two new rings are formed in a single synthetic operation, which proceeds through a metathesis reaction and attack of tethered carbon and heteroatom nucleophiles on iminium intermediates. The resulting bi-, tri-, and tetracylic ring systems are generally formed in good to excellent yields with excellent diasteroselectivities. We believe that our findings point in a promising direction for future metathesis research.

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- [1] For selected reviews on metathesis, see: a) T. Katz. Angew. Chem. 2005, 117, 3070-3079; Angew. Chem. Int. Ed. 2005, 44, 3010-3019; b) R. R. Schrock, J. Mol. Catal. A 2004, 213, 21-30; c) A. H. Hoveyda, R. R. Schrock, Compr. Asymmetric Catal. Suppl. 2004, 1, 207-233; d) R. R. Schrock, A. H. Hoveyda, Angew. Chem. 2003, 115, 4740-4782; Angew. Chem. Int. Ed. 2003, 42, 4592 – 4633; e) K. C. Nicolaou, P. G. Bulger, D. Sarlah, Angew. Chem. 2005, 117, 4564-4601; Angew. Chem. Int. Ed. 2005, 44, 4490-4527; f) P. H. Deshmukh, S. Blechert, Dalton Trans. 2007, 2479-2491; g) A. H. Hoveyda, A. R. Zhugralin, Nature 2007, 450, 243-251; h) G. C. Vougioukalakis, R. H. Grubbs, Chem. Rev. 2010, 110, 1746-1787; i) Handbook of Metathesis, Vol. 1-3 (Ed.: R. H. Grubbs), Wiley-VCH, Weinheim, 2003.
- [2] a) B. Alcaide, P. Almendros, A. Luna, Chem. Rev. 2009, 109, 3817-3858; b) B. Alcaide, P. Almendros, Chem. Eur. J. 2003, 9, 1258 - 1262
- [3] a) B. Schmidt, Eur. J. Org. Chem. 2004, 1865-1880; b) B. Schmidt, Pure Appl. Chem. 2006, 78, 469-476.
- [4] A. E. Sutton, B. A. Seigal, D. F. Finnegan, M. L. Snapper, J. Am. Chem. Soc. 2002, 124, 13390-13391.
- [5] a) B. Schmidt, J. Org. Chem. 2004, 69, 7672 7687; b) B. Schmidt, A. Biernat, Org. Lett. 2008, 10, 105-108.
- [6] For studies on isomerization during olefin metathesis, see: S. H. Hong, D. P. Sanders, C. W. Lee, R. H. Grubbs, J. Am. Chem. Soc. 2005, 127, 17160 – 17161, and references herein.
- [7] J. Louie, C. W. Bielawski, R. H. Grubbs, J. Am. Chem. Soc. 2001, 123, 11312 – 11313.
- [8] A. Fürstner, A. Leitner, Angew. Chem. 2003, 115, 320-323; Angew. Chem. Int. Ed. 2003, 42, 308-311.
- [9] C. Menozzi, P. I. Dalko, J. Cossy, Synlett 2005, 2449 2452.
- [10] B. G. Kim, M. L. Snapper, J. Am. Chem. Soc. 2006, 128, 52-53.
- [11] S. Beligny, S. Eibauer, S. Maechling, S. Blechert, Angew. Chem. 2006, 118, 1933-1937; Angew. Chem. Int. Ed. 2006, 45, 1900-1903.
- [12] A. A. Scholte, M. H. An, M. L. Snapper, Org. Lett. 2006, 8, 4759 - 4762.
- [13] a) B. A. Seigal, C. Fajardo, M. L. Snapper, J. Am. Chem. Soc. 2005, 127, 16329 – 16332; b) P. Quayle, D. Fengas, S. Richards,

- Synlett 2003, 1797 1800; c) C. D. Edlin, J. Faulkner, Tetrahedron Lett. 2006, 47, 1145-1151.
- [14] a) S. Fustero, M. Sánchez-Rosello, D. Jiménez, J. F. Sanz-Cervera, C. del Pozo, J. L. Aceña, J. Org. Chem. 2006, 71, 2706-2714; b) S. Fustero, E. Esteban, J. F. Sanz-Cervera, D. Jiménez, F. Mojarrada, Synthesis 2006, 4087-4091.
- [15] A. Mallagaray, G. Domínguez, A. Gradillas, J. Pérez-Castells, Org. Lett. 2008, 10, 597-600.
- [16] For selected reviews on N-acyliminium chemistry, see: a) W. N. Speckamp, H. Hiemstra, Tetrahedron 1985, 41, 4367-4416; b) W. N. Speckamp, M. J. Moolenaar, Tetrahedron 2000, 56, 3817-3856; c) B. E. Maryanoff, H. C. Zhang, J. H. Cohen, I. J. Turchi, C. A. Maryanoff, Chem. Rev. 2004, 104, 1431-1628; d) S. T. le Quement, R. Petersen, M. Meldal, T. E. Nielsen, Biopolymers 2010, 94, 242-256.
- [17] M. Martín-Martínez, N. De La Figuera, M. LaTorre, R. Herranz, M. T. García-López, E. Cenarruzabeitia, J. D. Río, R. González-Muñiz, J. Med. Chem. 2000, 43, 3770-3777.
- [18] D. Andreu, S. Ruiz, C. Carreño, J. Alsina, F. Albericio, M. A. Jiménez, N. de la Figuera, R. Herranz, M. T. García-López, R. González-Muñiz, J. Am. Chem. Soc. 1997, 119, 10579 – 10586.
- [19] T.-S. Kam, K.-M. Sim, *Phytochemistry* **1998**, 47, 145–147.
- [20] For selected noteworthy examples, see: a) T. E. Nielsen, M. Meldal, J. Org. Chem. 2004, 69, 3765-3773; b) I. T. Raheem, P. S. Thiara, E. A. Peterson, E. N. Jacobsen, J. Am. Chem. Soc. **2007**, 129, 13404–13405; c) T. Yang, L. Campbell, D. J. Dixon, J. Am. Chem. Soc. 2007, 129, 12070-12071; d) I. T. Raheem, P. S. Thiara, E. N. Jacobsen, Org. Lett. 2008, 10, 1577-1580; e) M. E. Muratore, C. A. Hooloway, A. W. Pilling, R. I. Storer, G. Trevitt, D. J. Dixon, J. Am. Chem. Soc. 2009, 131, 10796–10797; f) E. S. Sattely, S. J. Meek, S. J. Macolmson, R. R. Schrock, A. H. Hoveyda, J. Am. Chem. Soc. 2009, 131, 943-953; g) W. da Silva, M. T. Rodrigues, N. Shankaraiah, R. B. Ferreira, C. K. Z. Andrade, R. A. Pilli, L. S. Santos, Org. Lett. 2009, 11, 3238-3241; h) C. A. Holloway, M. E. Muratore, R. I. Storer, D. J. Dixon, Org. Lett. 2010, 12, 4720-4723.
- [21] For investigations into the instability of Δ^4 -pyrrolin-2-ones, see: J. T. Baker, S. Sifniades, J. Org. Chem. 1979, 44, 2798-2800.
- [22] S. H. Hong, R. H. Grubbs, Org. Lett. 2007, 9, 1955-1957.

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